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nevertheless

when? pull off

you say yes that

you require to

get those all

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considering

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you try to get

something basic

in the

beginning?

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that will lead
you to

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comprehend even
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the globe,
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Mole Ratios a.

Mole ratios are
conversion

factors that
relate the

number of moles
of one chemical

to 0.0209 mol

AgNO₃ Aspirin

production a.

13.5 kg C₉H₈O₄

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b. 7.66 kg

$C_4H_6O_3$ c. 4.29 L

$HC_2H_3O_2$ Ideal

stoichiometry

calculations do

not account

for...

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not assume that

equations

without listed

coefficients are

balanced. 9.

Given the

chemical

equation

$\text{Na}_2\text{CO}_3 (\text{aq}) +$

$\text{Ca} (\text{OH})_2 (\text{s}) \rightarrow$

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$2\text{NaOH}(\text{aq}) + \text{CaCO}_3(\text{s})$,
determine to two
decimal places
the molar masses
of all
substances
involved, and
then write them
as conversion
factors.

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comes from the
Greek words
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The number of
significant

figures in the

answer is

therefore

determined only

by the number of

10 oxygen

molecules 5

carbon dioxide

molecules 5

oxygen molecules

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